

## Explore Learning Calorimetry Lab Answers

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### Explore Learning Calorimetry Lab Answers

Calorimetry Lab. Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water.

### Calorimetry Lab Gizmo : Lesson Info : ExploreLearning

Answers To Explore Learning Calorimetry Student Exploration- Calorimetry Lab (ANSWER KEY)  
When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat Answers To Explore Learning Calorimetry Lab Gizmo

## **Answers To Explore Learning Calorimetry Lab Gizmo**

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## **Week\_1\_Online\_Lab\_Calorimetry\_Lab\_Gizmo.docx - Week 1 ...**

Student Exploration: Calorimetry Lab Vocabulary: calorie, calorimeter, joule, specific heat capacity  
Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. The Latin word calor means "heat," and meter comes from the Greek word meaning "to measure." What do you think a calorimeter does? Measures heat 2. Where have you heard the word calorie before?

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the initial temperature of 20 °C to result in a final system temperature of 20.45 °C. (Hint: Start by calculating how much heat energy is needed to change the water's temperature by 0.45 °C). Show your work. Use the Gizmo to check your answer.  $3000\text{g} \times 4.184 \text{ J/g} \times 0.45 \text{ }^\circ\text{C} = 5,648 \text{ J}$   $\times 0.795 \text{ J/g} \times 59.55 \text{ }^\circ\text{C} = 5,648 \text{ J}$  Mass of granite = 119.31g 5. Extend your thinking: In addition to calculating ...

## Suppose you place 125 g of aluminum in a calorimeter with ...

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